organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Rosica Petrova,^a* Boris Shivachev,^a Krasimir Kosev,^a Malinka Stoyanova^b and Silvia Angelova^b

^aBulgarian Academy of Sciences, CL of Mineralogy and Crystallography, Acad. G. Bonchev Str. build. 107, 1113 Sofia, Bulgaria, and ^bBulgarian Academy of Sciences, Institute of Organic Chemistry, Acad. G. Bonchev Str. build. 9, 1113 Sofia, Bulgaria

Correspondence e-mail: rosica.pn@clmc.bas.bg

Key indicators

Single-crystal X-ray study T = 290 KMean σ (C–C) = 0.010 Å R factor = 0.055 wR factor = 0.129 Data-to-parameter ratio = 7.8

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

© 2005 International Union of Crystallography Printed in Great Britain – all rights reserved

(3*RS*,4*RS*)-3-(2-Furyl)-2-phenethyl-4-(pyrrolidin-1-yl-carbonyl)-3,4-dihydroisoquinolin-1(2*H*)-one

The title compound, $C_{26}H_{26}N_2O_3$, crystallizes as a racemic mixture of *R*,*R* and *S*,*S* enantiomers. The two independent molecules show a similar conformation. They are connected *via* C-H···O and C-H··· π interactions to form ribbons along the *a* axis.

Received 10 June 2005 Accepted 16 June 2005 Online 24 June 2005

Comment

2,3,4-trisubstituted tetrahydroisoquinolines are important from both a technological and an applied point of view as part of a significant class of compounds with a variety of pharmacological applications (Jayaraman *et al.*, 2000; Zhang *et al.*, 1997; Bernan *et al.*, 1994; Koleva *et al.*, 1998). In the course of our research on the crystal structure and biological activity of heterocyclic molecules, the title compound, (I), was synthesized. Although the presence of pyrrolidine and furyl substituents suggests biological activity of this compound it has not been confirmed experimentally. In this paper, we describe the crystal structure of (I).



There are two independent molecules, A and B in the asymmetric unit of (I) (Fig. 1). The chemical scheme corresponds to the R,R isomer. All intramolecular bond distances and angles are within expected ranges (Gzella *et al.*, 2002; Georgieva *et al.*, 1994). In molecules A and B, the aromatic and furyl rings are nearly planar, the pyridinone rings adopt a twist conformation, and the pyrrolidine rings show an envelope conformation (Table 1). At the same time, according to IUPAC terminology (Moss, 1996), both furyl and pyrrolidine groups are in pseudo-axial positions with respect to the dihydroisoquinoline group.

Molecules A and B participate in similar three-dimensional molecular packing interactions (Fig. 2). Symmetry-equivalent molecules are linked *via* weak $C-H\cdots O$ interactions (Table 2) between one of the furyl C atoms and the dihydro-isoquinolinone O atom, forming zigzag chains along the *a* axis.



Figure 1

The structures of the two independent molecules, A and B, in (I), showing 50% probability displacement ellipsoids.



Figure 2

The molecular packing, with *B* molecules colored in blue. The dotted lines indicate $C-H\cdots\pi$ and $C-H\cdots$ O interactions. [Symmetry codes: (i) $x - \frac{1}{2}, 2 - y, z$; (ii) $\frac{1}{2} + x, 1 - y, z$; (iii) x - 1, y, z; (iv) 1 + x, y, z.]

Molecules A and B are connected via intermolecular C– $H_{furyl} \cdot \cdot \pi$ interactions (Table 2) to form almost-centrosymmetric dimer units. The angles between the normal to the plane of the aromatic unit and the line linking the C atom and the centroid of the ring are 22.1 (2) and 21.0 (1)° in molecules A and B, respectively.

Experimental

Compound (I) was synthesized by the well known reaction between homophthalic anhydride and imine (Haimova *et al.*, 1977; Kozekov *et al.*, 2002; Stoyanova *et al.*, 2003), leading to the corresponding *trans*and *cis*-carboxylic acids and their subsequent transformations by analogy to Haimova *et al.* (1984). The product was characterized by IR, ¹H and ¹³C NMR spectra. Crystals of (I) were obtained by slow evaporation (about five months) of a dimethylformamide solution.

Crystal data

 $C_{26}H_{26}N_2O_3$ $M_r = 414.49$ Orthorhombic, *Pca2*₁ a = 11.9675 (15) Å b = 11.066 (2) Å c = 32.7891 (14) Å $V = 4342.3 (10) \text{ Å}^3$ Z = 8 $D_r = 1.268 \text{ Mg m}^{-3}$

Data collection

Enraf-Nonius CAD-4 diffractometer non-profiled $\omega/2\theta$ scans Absorption correction: none 8655 measured reflections 4339 independent reflections 2189 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.076$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.055$ $wR(F^2) = 0.129$ S = 0.984339 reflections 559 parameters

reflections $\theta = 16.1-17.3^{\circ}$ $\mu = 0.08 \text{ mm}^{-1}$ T = 290 (2) KPrism, yellow $0.26 \times 0.2 \times 0.2 \text{ mm}$

Cell parameters from 22

Mo $K\alpha$ radiation

 $\begin{array}{l} \theta_{\max} = 26.0^{\circ} \\ h = 0 \rightarrow 14 \\ k = 0 \rightarrow 13 \\ l = -40 \rightarrow 40 \\ 3 \text{ standard reflections} \\ \text{frequency: } 120 \text{ min} \\ \text{intensity decay: } 1\% \end{array}$

H-atom parameters constrained
$w = 1/[\sigma^2(F_o^2) + (0.0463P)^2]$
where $P = (F_0^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{\rm max} < 0.001$
$\Delta \rho_{\rm max} = 0.16 \ {\rm e} \ {\rm \AA}^{-3}$
$\Delta \rho_{\rm min} = -0.18 \text{ e } \text{\AA}^{-3}$

Table 1

Selected torsion angles (°).

31.5 (8)	C24-C23-N22-C21	-36.8(7)
-46.2(6)	N22-C23-C24-C25	49.4 (7)
33.9 (7)	C23-C24-C25-C26	-34.0(7)
-36.0(10)	N21-C224-C225-C226	-28.4(8)
28.4 (9)	C224-C225-C226-C227	33.2 (8)
	31.5 (8) -46.2 (6) 33.9 (7) -36.0 (10) 28.4 (9)	31.5 (8) C24-C23-N22-C21 -46.2 (6) N22-C23-C24-C25 33.9 (7) C23-C24-C25-C26 -36.0 (10) N21-C224-C25-C226 28.4 (9) C224-C25-C226-C227

Table 2 Hydrogen-bo

Hydrogen-bond geometry (Å, $^{\circ}$).

Cg1 and Cg2 are the centroids of the dihydroisoquinoline aromatic rings of molecules A and B, respectively.

	D II	TT 4	D 4	
$D - H \cdots A$	D-H	$\mathbf{H} \cdots \mathbf{A}$	$D \cdots A$	$D - H \cdots A$
C122-H122···O11 ⁱ	0.93	2.49	3.342 (8)	153
$C222 - H222 \cdot \cdot \cdot O21^{ii}$	0.93	2.57	3.454 (8)	160
$C121 - H121 \cdots Cg2^{iii}$	0.93	2.89	3.681 (6)	147
$C221 - H221 \cdots Cg1^{iv}$	0.93	2.92	3.756 (5)	149
	1	(**) · 1		1 (1)

Symmetry codes: (i) $x - \frac{1}{2}, -y + 2, z$; (ii) $x + \frac{1}{2}, -y + 1, z$; (iii) x - 1, y, z; (iv) x + 1, y, z.

The H atoms were placed in idealized positions (C–H = 0.93–0.97 Å) and were constrained to ride on their parent atoms, with $U_{\rm iso}({\rm H}) = 1.2 U_{\rm eq}({\rm C})$. Friedel-pair reflections were merged, since anormalous scattering effects were negligible.

Data collection: *CAD-4 EXPRESS* (Enraf–Nonius, 1994); cell refinement: *CAD-4 EXPRESS*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP3 for Windows* (Farrugia, 1997) *MERCURY* (Version 1.3; Bruno *et al.*, 2002); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

References

- Bernan, V. S., Montenegro, D. A., Korshalla, J. D., Maiese, W. M., Steinberg, D. A. & Greenstein, M. (1994). J. Antibiot. 47, 1417–1423.
- Bruno, I. J., Cole, J. C., Edgington, P. R., Kessler, M. K., Macrae, C. F., McCabe, P., Pearson, J. & Taylor, R. (2002). Acta Cryst. B58, 389–397.
- Enraf-Nonius (1994). CAD-4 EXPRESS. Enraf-Nonius, Delft, The Netherlands.
- Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
- Farrugia, L. J. (1999). J. Appl. Cryst. 32, 837-838.

- Georgieva, A., Stanoeva, E., Karamfilova, K., Spassov, S., Angelova, O., Haimova, M., De Kimpe, N. & Boelens, M. (1994). *Tetrahedron*, **50**, 9399– 9410.
- Gzella, A., Brozda, D., Koroniak, L. & Rozwadowska, M. (2002). *Acta Cryst.* C58, o503–o506.
- Haimova, M., Atanasova, I., Stanoeva, E. & Mihovska, S. (1984). Commun. Dept Chem. Bulg. Acad. Sci. 17, 163–171.
- Haimova, M. A., Mollov, N. M., Ivanova, S. C., Dimitrova, A. I. & Ognyanov, V. I. (1977). *Tetrahedron*, **33**, 331–336.
- Harms, K. & Wocadlo, S. (1995). XCAD4. University of Marburg, Germany. Jayaraman, M., Fox, B. M., Hollingshead, M., Kohlhagen, G., Pommier, Y. &
- Cushman, M. (2000). J. Med. Chem. 43, 3688–3695.
- Koleva, M., Stanoeva, E., Stoychev, Ts. & Haimova, M. (1998). Ann. Univ. de Sofia Fac. Chem. 90, 211–219.
- Kozekov, I. D., Koleva, R. I. & Palamareva, M. D. (2002). J. Heterocycl. Chem. **39**, 229–236.
- Moss, G. P. (1996). Pure Appl. Chem. 68, 2193-2222.
- Sheldrick, G. M. (1997). SHELXL97 and SHELXS97. University of Göttingen, Germany.
- Stoyanova, M. P., Kozekov, I. D. & Palamareva, M. D. (2003). J. Heterocycl. Chem. 40, 795–803.
- Zhang, H., Zembower, D. & Chen, Z. (1997). Bioorg. Med. Chem. Lett. 7, 2687–2690.